

A MULTI-MODEL APPROACH FOR THE MONITORING OF CARBON AND NITROGEN CONCENTRATIONS DURING THE AEROBIC PHASE OF A BIOLOGICAL SEQUENCING BATCH REACTOR

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Abstract: This paper presents i) an extension of a simple model describing the dynamical behavior of important variables during the aerobic phase of a biological wastewater treatment plant and ii) a simple asymptotic observer based on this extended model for on-line estimating both soluble biodegradable carbon and nitrate+nitrite nitrogen concentrations. The proposed model is identified and the observer is validated with real data acquired on a real biological Sequencing Batch Reactor pilot plant.
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1. INTRODUCTION

Activated sludge Sequencing Batch Reactor (SBR) is a reliable and low cost treatment process. But most importantly, its advantages are related to its ability to easily deal with sludge settlement which is difficult to maintain in good conditions in classical activated sludge processes due to the persistent incoming hydraulic shock loads

The SBR process is characterized by a series of process phases : fill, react, settle and draw taking place in the same tank. Initially, the reactor contains a volume V_0 and a quantity of biomass X_0 that remains from the last cycle. The cycle starts by introducing a volume $\Delta V = V_{\max} - V_0$ into the reactor with a flow rate Q . The reaction phase can be divided in two sub-phases : aerated and non-aerated. These two steps

allow both carbon and nitrogen removal. During these phases the reactor is maintained perfectly homogeneous. Once the reaction phase is completed, the agitation is stopped and the sludge starts flocculating and settling. The clean supernatant is then separated from the sludge and can be withdrawn from the reactor. The reactor is then available to receive a new volume ΔV of wastewater. After few cycles, because the pollution is transformed into biomass, it is necessary to waste some sludge (Wilderer *et al.*, 2001).

A reduced and simple model for control design and optimization of SBR activated sludge processes was proposed in (Mazouni *et al.*, 2004). The model takes into account the three more important biological phenomena : aerobic carbon and ammonium nitrogen oxydation and anoxic carbon and nitrate

denitrification. The proposed model was identified and cross-validated using 15 experiments that were carried out from 05.08.2003 to 15.04.2004 in the Laboratory of Environmental Biotechnology (LBE) INRA Narbonne, France.

One of the main difficulties in control design and its practical implementation is the lack of cheap and reliable on-line sensors for the main process variables. To overcome this problem an asymptotic observer of both soluble biodegradable carbon, S_1 , and nitrate+nitrite nitrogen concentrations, S_3 during the aerobic phase is proposed hereafter (Bastin et al 1990). In order to take advantage of the information contained in the dissolved oxygen concentration (which is a cheap and reliable industrial sensor), an extended model including its dynamics is used here instead of the simple model proposed in (Mazouni *et al.*, 2004). Then, the observer is derived on the basis of this extended model.

The model parameters are identified and the observer is cross-validated. For this purpose, 6 experiments that were carried out from 27.10.2003 to 12.11.2004 in the LBE are used. More specifically, the experiments are classified in three groups depending of the value of oxygen mass transfer coefficient, $K_L a$. One set of model parameters is identified for each group. Finally, some conclusions and perspectives are drawn

2. MODELLING

A schematic view of the process available at the LBE, Narbonne, France, is given in figure 1.

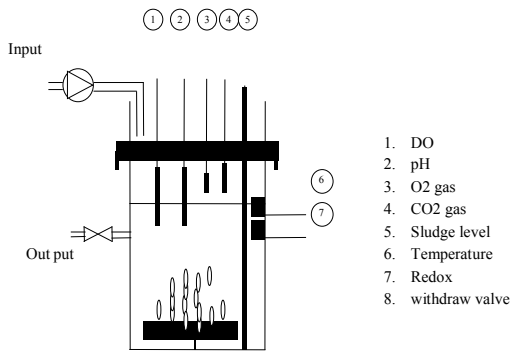


Fig 1 : SBR pilot plant

The tank of the reactor has a cylindrical form. Its dimensions are 50cm \varnothing and 130cm height with a total volume of 255L. It is equipped with a variable flow rate pump to fill the reactor and a controlled valve to withdraw the effluent and the sludge in excess. Air is used for aeration and mixing. Its flow is controlled by a flow meter.

The reactor is operated at an ambient lab temperature of 20°C. Each cycle duration is 24 hours. More specifically, the reaction phases are divided into 2 periods. In each one, half of the total influent daily

volume is added (25 liters). A schematic view of the operating modes is shown in figure 2.

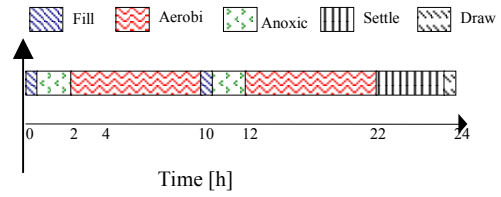


Fig.2 SBR cycle phases

Because both the carbon and nitrogen are treated, two specific phases are needed : aerobic (presence of oxygen) and anoxic (no oxygen but nitrite and nitrate). During the anoxic phase, the nitrite+nitrate from the last cycle are transformed into gaseous nitrogen. This step is realized by heterotrophic bacteria who also need soluble organic carbon to grow (Hu et al, 2003)(Jeppsson, 1996). When all the nitrite-nitrate has been removed, one can switch to aerobic mode where the ammonium nitrogen will be transformed into nitrite and nitrate while the remaining soluble carbon will be removed.

From now, we only concentrate on what happens during the aerobic phase and the following notations are used :

X_1 : Heterotrophic microorganisms. (mg/l)

X_2 : Autotrophic microorganisms (mg/l)

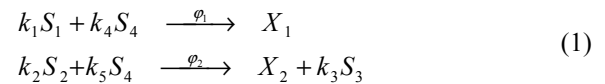
S_1 : Organic carbon. (mg COD /l)

S_2 : Ammonium nitrogen. (mg N-NH4/l)

S_3 : Nitrates/Nitrites nitrogen (mg N-NO3/l+ mg N-NO2/l)

S_4 : Dissolved oxygen (mg O2/l)

The reaction scheme of the aerobic phase is as follows :



This synthetic scheme means that S_1 and S_2 are degraded by X_1 and X_2 respectively with the rates φ_1 and φ_2 in the presence of S_4 . In other words, the pollutant S_1 serves as a basis for the growth of X_1 while S_2 is transformed into S_3 by X_2 .

Considering the above reaction scheme, we can apply the mass balance principle to determine the state space model of the aerobic phase (Bastin et al 1990):

$$\dot{X}_1 = \mu_1(S_1, S_4) X_1 \quad (2)$$

$$\dot{X}_2 = \mu_2(S_1, S_4) X_2 \quad (3)$$

$$\dot{S}_1 = -k_1 \mu_1(S_1, S_4) X_1 \quad (4)$$

$$\dot{S}_2 = -k_2 \mu_2(S_1, S_4) X_2 \quad (5)$$

$$\dot{S}_3 = k_3 \mu_2(S_1, S_4) X_2 \quad (6)$$

$$\dot{S}_4 = -k_4\mu_1(S_1, S_4)X_1 - k_5\mu_2(S_1, S_4)X_2 + K_L a(S_4^{\max} - S_4)$$

with

$$\begin{aligned}\mu_1 &= \mu_{1\max} \frac{S_1 - S_1^*}{K_{S1} + S_1 - S_1^*} S_4; \\ \mu_2 &= \mu_{2\max} \frac{S_2}{K_{S2} + S_2} S_4.\end{aligned}\quad (8)$$

As can be seen, the specific growth rates in aerobic phase, μ_1 and μ_2 , are proportional to the dissolved oxygen concentration, S_4 , in the reactor instead of the classical expression $\frac{S_4}{K_o + S_4}$ proposed in (Henze *et al.*, 1987). It is essentially because the available experiments were realized with a non-limiting oxygen concentration, S^* being the average of residual COD in the reactor (Mazouni *et al.*, 2004).

3. MODEL IDENTIFICATION

It was shown in (Mazouni *et al.*, 2004) that the model is very sensitive to the oxygen transfer coefficient $K_L a$. This parameter is itself very sensitive to environmental conditions such as input air flow rate. When operating in real conditions, it is possible to classify the operating conditions into a finite number of classes depending on the input air flow rate. That way, a limited number of set of process parameters can be identified and, depending on the environmental conditions, the user can predict the future behavior of the process in using the specific model associated with the actual operating conditions.

Table 1 Influent concentration and air flow rate for each experimental data used for identification

Exp. No	Air flow (l/min)	COD in (mg/l)	NTK in (mg/l)	$K_L a$ (h^{-1})
27.10.03	50	4318.18	168.00	18.75
29.10.03	50	4772.25	183.68	18.75
31.10.03	50	4476.92	204.40	18.75
04.11.03	50	4836.85	319.20	18.75
06.11.03	75	4235.45	260.4	23.00
12.11.03	25	4945.61	285.60	9.60

Remember that the aerobic model consists of six variables and nine parameters. The parameters $\mu_{\max 1}$ and $\mu_{\max 2}$ are the maximum specific growth rates of X_1 and X_2 (Heterotrophic and Autotrophic microorganisms respectively); K_{S1} and K_{S2} are saturation constants associated with substrates S_1 and S_2 respectively; k_1 , k_2 and k_3 are the yield coefficients for degradation of S_1 by X_1 and S_2 by X_2 ; and for the production of S_3 by X_2 respectively. The coefficients k_4 and k_5 are oxygen yield coefficients associated with X_1 and X_2 respectively while S_4^{\max} is the maximal value of dissolved oxygen in the liquid medium.

In other words, we use a multi-model approach : the process is not described by only one model but by n models, each of them being related to a given range of input air flow rates (a value of $K_L a$).

For the parameter identification, the procedure used is the same than the one proposed in (Mazouni *et al.*, 2004). Available experimental data correspond to 6 aerobic SBR cycles where the influent concentrations and the input air flow rate were changed as mentioned in Table 1. Four experiments were performed with an influent concentration of 50 l/min (namely the experiments realized on 27.10.03, 29.10.03, 31.10.03 and 04.11.03). The influent concentration for the experiment No. 06.11.03 is 75 l/min and for experiment No. 12.11.03 is 25 l/min. In the last column in Table 1, the values of oxygen mass transfer coefficients, $K_L a$, that are calculated for each experiment are shown. As can be seen, three types (classes) of experiments could be distinguished, according the value of $K_L a$. The first one consists of four experiments with $K_L a=18.75$; the two others listed in the Table 1 consist of one experiment namely 06.11.03 and 12.11.03 with $K_L a=23.00$ and $K_L a=9.60$ respectively.

The experiments No. 27.10.03, 29.10.03, 31.10.03 are used for parameter identification of process model of the group with $K_L a=18.75$. The experiment 04.11.03 is used for cross-validation of this model. The others two groups are presented with an unique experiment for each group. The experiments were carried out in winter periods of the year : three experiments in October, and three in November.

An optimization algorithm that was proposed in (Mazouni *et al.*, 2004) is applied for model parameter estimation. The optimization criterion is chosen to obtain the minimal error between experimental data and simulated outputs of the system (2)-(8). The results of parameter estimation are shown in Table 2. As can be seen, different sets of parameter values are obtained for each class of experiments depending on the value of $K_L a$. Model parameter values of the group with $K_L a=18.75$ are obtained as medium values of three experiments and cross-validated with the data of the fourth one. The model parameters are shown in the Table 2.

Table 2 Estimated parameters for aerobic model

Exp. N	μ_{1max}	μ_{2max}	K_{S1}	K_{S2}	k_1	k_2	k_3	k_4	k_5
27.10.03	0.014	0.023	192.7	93.3	5.81	0.55	0.25	0.35	8.96
29.10.03	0.023	0.016	107.9	91.8	1.99	0.76	0.32	2.43	7.25
31.10.03	0.031	0.013	112.3	101.7	2.34	0.71	0.27	2.43	7.33
model	0.023	0.017	137.6	95.6	3.38	0.67	0.28	1.74	7.85
06.11.03	0.027	0.012	109.2	111.9	2.63	0.64	0.30	1.99	6.99
12.11.03	0.031	0.013	98.2	101.3	4.67	0.90	0.28	1.40	6.02

In Figures 3 (a,b,c and d), the model for experiments with $K_L a = 18.75$ is cross-validated with data from the experiment No. 04.11.03 (not listed in Table 2). The model parameters are obtained as medium values of

parameters of the other experiments of the same group that are used for model identification.

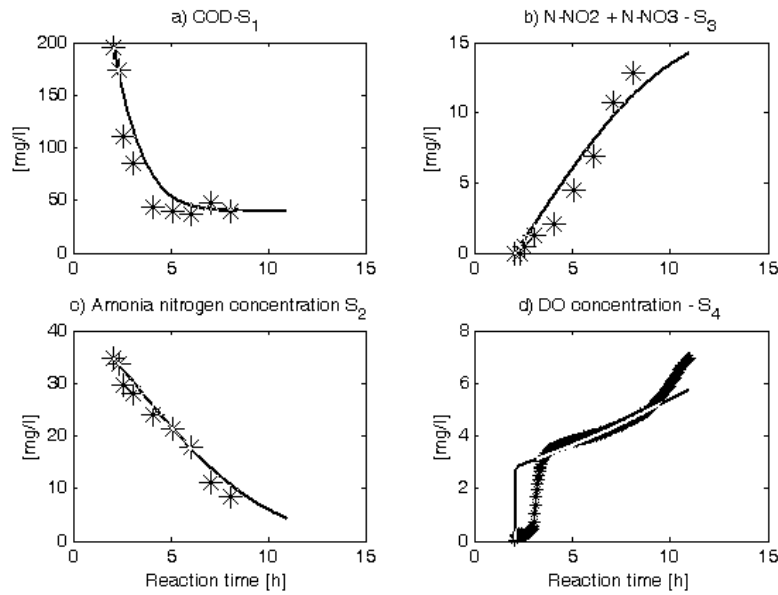


Fig.4. Model parameter cross-validation. (-)model and No 04.11.03 experimental data (*).

4. OBSERVER DERIVATION

In the last section, we have derived three models for three different values of input air flow rates. The model parameters have been estimated. It should be noticed that these models could be used to derive observers of diverse classes. However, biological processes exhibit uncertain behaviours and our objective is to design a system able to monitor S_1 and S_3 from the on-line measurements of S_2 and S_4 – independently of the reaction kinetics – which, with no doubt, are the terms that are the most affected by uncertainty.

The process model (equations (2)-(8)) can be written in a matrix format as follows:

$$\dot{\xi} = K\varphi + F \quad (9)$$

where $\xi = [S_1 \ S_2 \ S_3 \ S_4 \ X_1 \ X_2]^T$ is the vector of process variables; K is the matrix of yield coefficients; $\varphi = [\varphi_1 \ \varphi_2]^T$ is the vector of reaction rates and F is the vector of mass feed rate in the reactor.

The model (9) describes the batch aerobic autotrophic growth and consists of two main parts. The first term represents process kinetics and the second one represents the transport dynamics. From now, the process kinetics is only partially known : the reaction rates ($\varphi_1 = \mu_1 X_1$ and $\varphi_2 = \mu_2 X_2$) are unknown while the values of the yield coefficients in the matrix K are known. The transport dynamics is known. It consists of oxygen supply by aeration.

Introducing the auxiliary variables :

$$\begin{cases} Z_1 = S_2 + \frac{k_2}{k_3} S_3 \\ Z_2 = S_4 - \frac{k_4}{k_1} S_1 + \frac{k_5}{k_3} S_2 \end{cases} \quad (10)$$

allows us to consider the following dynamical sub-model :

$$\begin{cases} \frac{dZ_1}{dt} = 0 \\ \frac{dZ_2}{dt} = K_L a (S_4^{\max} - S_4) \end{cases} \quad (11)$$

Assuming that S_2 and S_4 are measured on-line and that $Z_1(0) = Z_{10}$ and $Z_2(0) = Z_{20}$ are known, the system (10)-(11) can be used to get an on-line estimation of S_1 and S_3 as :

$$\begin{cases} \hat{S}_1 = \frac{k_5 k_1}{k_2 k_4} (Z_1 - S_2) - \frac{k_1}{k_4} (Z_2 - S_4) \\ \hat{S}_3 = \frac{k_3}{k_2} (Z_1 - S_2) \end{cases} \quad (12)$$

Notice that equations (12) is not an "observer" (nor even a detector!) since it is necessary to exactly know both Z_{10} and Z_{20} to correctly estimate S_1 and S_3 from the on-line measurements of S_2 and S_4 . Rather it is a tricky way of estimating S_1 and S_3 without getting any idea of μ_1 and μ_2 .

6. RESULTS AND DISCUSSION

The multi-model monitoring system for the first group was validated by comparing the observation curves with data of experiment No. 04.11.03. The comparison between the real data and the output of the monitoring system are shown in figures 4. The necessity of determining the initial values of the system (11) can be solved as follows. As explained in the introduction, at the beginning of the aerobic phase, there is no nitrite nor nitrate in the reactor. Thus $S_3(0) = 0$. Since S_2 is measured, $S_2(0) = S_{20}$ is known. From equation (10), one concludes that $Z_1(0) = k_3 S_{20} + k_2 S_{30}$. Since S_4 is also measured one has only to assume that the initial value of the organic carbon to be removed in one cycle is known ($S_1(0) = S_{10}$) in order to compute $Z_2(0)$ using equation (10b) as $Z_2(0) = S_{40} - k_4 S_{10} / k_1 + k_5 S_{20} / k_3$.

Usually, at the beginning of the process, the initial value of COD (or at least an average value of it) is known (measured off-line). The values of COD_{in} for the experiments used in this paper are shown in Table 1. It is observed that the initial value of COD at the beginning of the aerobic phase is between 3% and 4% of the value of COD_{in} listed in Table 1. The simulation in Figure 3a is run with 3.5% of COD_{in} of the experiment No. 04.11.03.

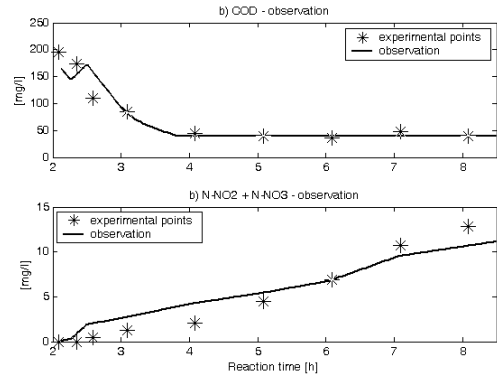


Fig. 4. Experimental No 04.11.03 data and estimation of S_1 and S_2

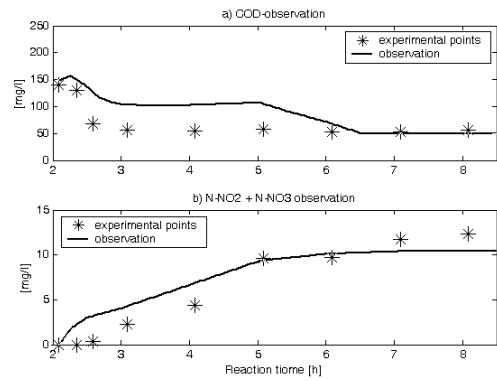


Fig. 5. Experiment No 06.11.03 data and estimation of S_1 and S_2

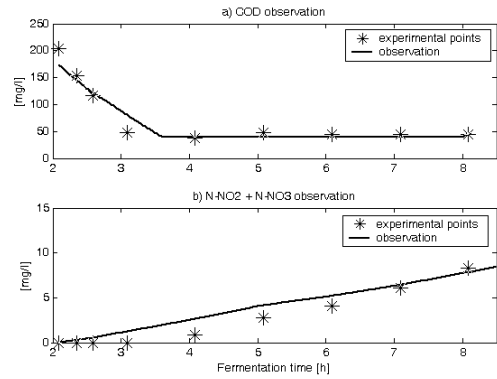


Fig. 6. Experiment No 12.11.03 data and estimation of S_1 and S_3

As can be seen in Figure 4a, observations fits well with off-line measured points until 4h30. After that time, a saturation of the COD observed value is included in the observation algorithm to overcome the difference between the data and the monitoring system. That difference relies on the fact that COD measurement includes a part of non-biodegradable COD (usually self generated by biomass decay (Orhoni *et al.*, 2003)). This residual COD is a constant value and is around 40 mg/l

In Figures 5 and 6, the observation concerning the others two groups are validated with the experimental points.

7 CONCLUSIONS

A simple multi-model monitoring system for both on-line estimating soluble biodegradable carbon and nitrate+nitrite nitrogen concentrations during the aerobic phase of an SBR process was presented. A full model – including the dynamics of the dissolved oxygen - is identified and validated with real data acquired on a SBR pilot plant operating in the LBE, Narbonne, France. The experiments were performed in October – November 2003. They are classified in three groups according to the values of $K_L a$. Three experiments with $K_L a=18.75$ are used for model identification of this group while additional data are used for model validation. A sub-model –independent from the reaction kinetics – is derived and a multi-model monitoring system allowing the on-line estimation of both the soluble organic carbon and the nitrite+nitrate concentrations is proposed. The observation of nitrogen fits well with experimental points while the differences observed between the estimation and the real COD data could be overcome by a saturation included in the observation algorithm. Such a saturation is accepted because of the presence of non biodegradable COD generated by the biomass decay.

8. ACKNOWLEDGEMENT

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