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EXAM IN SUBJECT TEP4170 HEAT AND COMBUSTION TECHNOLOGY (Varme- og forbrenningsteknikk) 21 May 2012 Time: 0900 – 1300

The exam is only available in English. The answers can be written in Norwegian or English.

Permitted aids: D – No printed or handwritten aids. Certain simple calculator.

- Please do not use red pencil/pen, as this is reserved for the censors.
- Read through the problems first. Begin with the problem where you feel that you have the best insight. If possible, do not leave any problems blank. <u>Formulate clearly</u>, it pays off!

NOTE: The decimal sign is <u>comma</u>.

Problems:

1)

--Explain what is meant by "global" reactions and "elementary" reactions.

--What is meant by "unimolar", "bimolar" and "trimolar" in this context?

--What is meant by "reaction order" (in this context)? What distinguishes global and elementary reactions in this respect?

2)

Given the following set of three reaction pairs

$$H + O_{2} \xrightarrow{k_{1f}} OH + O$$

$$OH + O \xrightarrow{k_{1f}} H + O_{2}$$

$$O + H_{2} \xrightarrow{k_{2f}} OH + H$$

$$OH + H \xrightarrow{k_{2f}} O + H_{2}$$

$$OH + H_{2} \xrightarrow{k_{3f}} H_{2}O + H$$

$$H_{2}O + H \xrightarrow{k_{3r}} OH + H_{2}$$

--Each pair can be assumed to be in equilibrium (Norw: "jamvekt"). Express the molar concentration of the radical species O, H and OH in terms of the kinetic rate coefficients (Norw: "reaksjonskoeffisientar") and the molar concentrations of H₂, O₂ and H₂O. --What are the conditions for the assumption that the three pairs of reactions are in equilibrium, and when will this be satisfied?

3)

-- Define isotropic turbulence.

--The turbulence equations can be simplified for isotropic turbulence. Put up the equations for turbulence energy (k) and its dissipation rate (ε).

--Will k and ε increase or decrease? Explain!

--Put up expressions for all the Reynolds stresses (turbulence stresses), $-\rho \overline{u'_i u'_j}$, for isotropic turbulence, assuming that k and ε are known.

4)

-- Describe the principle for determining the flammability limits (Norw:"flammegrenser").

-- How are flammability limits specified? (i.e. in which form are the values given)

5)

The axes of the Borgi diagaram are a length-scale ratio (horizontal axis) and a velocity-scale ratio (vertical axis).

--Sketch the Borghi diagram.

--Include lines for the turbulence Reynolds number, for the Damköhler number, and for the small-scale Damköhler number (i.e. based on the Kolmogorov microscales). --Define the quantities used.

6)

-- Identify different regimes of non-premixed flames in the Borghi diagram.

-- Locate a (conventional) wood stove, a (conventional, non-premixed) gas turbine, a large furnace/boiler, and a large open-air fire in the sketched diagram. Explain the locations.

7)

In a part of the boundary layer, the following approximate relation is found: $\frac{\partial \overline{u}_1}{\partial x_2} = \frac{u_{\tau}}{\kappa x_2}$

Use this expression, and (if required) other approximations/assumptions to develop wall functions (Norw: "vegglover") for the non-dimensional velocity and for the dissipation rate of turbulence energy.

8)

The transport equation for the variance of the mixture fraction (Norw.: "blandingsfraksjon"), ξ , can be written (assuming constant density) as

$$\frac{\partial}{\partial t}(\rho\overline{\xi'}^2) + \frac{\partial}{\partial x_j}(\rho\overline{\xi'}^2\overline{u}_j) = \frac{\partial}{\partial x_j}\left(D\frac{\partial\overline{\xi'}^2}{\partial x_j} - \rho\overline{\xi'}^2u_j'\right) - 2\rho\overline{\xi'}u_j'\frac{\partial\overline{\xi}}{\partial x_j} - 2D\frac{\overline{\partial\xi'}}{\partial x_j}\frac{\partial\xi'}{\partial x_j}$$

--Write the transport equation for the instantaneous mixture fraction, on which the equation above is based.

-- Show the development of the variance equation above (the principle, all terms of all intermediate equations are not required).

9)

Air is flowing in a straight duct (channel) with a square cross-section. The width of the duct is 0,1 m, and the volumetric flow is 0,1 m³/s. The viscosity can be assumed at $v = 1.5 \cdot 10^{-5} \text{ m}^2/\text{s}$.

--Verify that this flow is turbulent, and estimate values for the turbulence energy, k, the dissipation rate of turbulence energy, ε , the Kolmogorov length scale, η , and the turbulence Reynolds number.

10)

Sketch a mass-time diagram appearing from a thermo-gravimetric analysis (TGA) of a biomass fuel sample. Describe the stages seen in the diagram.